according to 29CFR1910/1200 and GHS Rev. 3

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#### Calcium Metal,

## SECTION 1: Identification of the substance/mixture and of the supplier

Product name: Calcium Metal,

Manufacturer/Supplier Trade name:

Manufacturer/Supplier Article number: \$25217

Recommended uses of the product and uses restrictions on use:

**Manufacturer Details:** 

AquaPhoenix Scientific 9 Barnhart Drive, Hanover, PA 17331

## **Supplier Details:**

Fisher Science Education 15 Jet View Drive, Rochester, NY 14624

## **Emergency telephone number:**

Fisher Science Education Emergency Telephone No.: 800-535-5053

## **SECTION 2: Hazards identification**

# Classification of the substance or mixture:



#### **Flammable**

Substances and mixtures, which in contact with water, emit flammable gases, category 2

Hazards Not Otherwise Classified - Combustible Dust Substances and mixtures which in contact with water emit flammable gases - Water-react. 2 Hazards Not Otherwise Classified - Water Reactive

Signal word: Danger

#### **Hazard statements:**

In contact with water releases flammable gas

#### **Precautionary statements:**

If medical advice is needed, have product container or label at hand

Keep out of reach of children

Read label before use

Do not eat, drink or smoke when using this product

Keep away from any possible contact with water, because of violent reaction and possible flash fire

Handle under inert gas. Protect from moisture

Wear protective gloves/protective clothing/eye protection/face protection

Brush off loose particles from skin. Immerse in cool water/wrap in wet bandages

In case of fire: Use agents recommended in section 5 for extinction

Store in a dry place. Store in a closed container

Dispose of contents and container as instructed in Section 13

#### **Water Reactive:**

Reacts violently and/or explosively with water, steam or moisture; dangerous when wet

#### **Combustible Dust Hazard:**

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#### Calcium Metal,

Can be explosive when exposed to heat or flames; may form combustible dust (explosive) concentrations in air (during processing).

#### Other Non-GHS Classification:







#### NFPA/HMIS





NFPA SCALE (0-4)

HMIS RATINGS (0-4)

# **SECTION 3: Composition/information on ingredients**

Ingredients:			
CAS 7440-70-2	Calcium	100 %	
		Percentages are by weight	

## **SECTION 4: First aid measures**

## **Description of first aid measures**

After inhalation: Loosen clothing as necessary and position individual in a comfortable position. Move exposed to fresh air. Give artificial respiration if necessary. If breathing is difficult give oxygen. Get medical assistance if cough or other symptoms appear.

After skin contact: Rinse/flush exposed skin gently using soap and water for 15-20 minutes. Seek medical advice if discomfort or irritation persists.

**After eye contact:** Protect unexposed eye. Rinse/flush exposed eye(s) gently using water for 15-20 minutes. Remove contact lens(es) if able to do so during rinsing. Seek medical attention if irritation persists or if concerned.

After swallowing: Rinse mouth thoroughly. Do not induce vomiting. Have exposed individual drink sips of water. Seek medical attention if irritation, discomfort or vomiting persists. Never give anything by mouth to an unconscious person.

## Most important symptoms and effects, both acute and delayed:

Irritation, Nausea, Headache, Shortness of breath. Irritation/burns, all routes of exposure. Contact with skin causes irritation and possible burns, especially if the skin is wet or moist. Causes eye irritation and possible burns;

#### Indication of any immediate medical attention and special treatment needed:

according to 29CFR1910/1200 and GHS Rev. 3

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#### Calcium Metal,

If seeking medical attention, provide SDS document to physician. Physician should treat symptomatically.

## **SECTION 5: Firefighting measures**

## **Extinguishing media**

## Suitable extinguishing agents:

For safety reasons unsuitable extinguishing agents: Do NOT use water

## Special hazards arising from the substance or mixture:

Thermal decomposition can lead to release of irritating gases and vapors. Water Reactive; material will react with water and may release a flammable and/or toxic gas. This chemical poses an explosion hazard - reacts with water to form explosive hydrogen gas. May ignite or explode on contact with steam or moist air; may reignite after fire is extinguished

## **Advice for firefighters:**

**Protective equipment:** Use NIOSH-approved respiratory protection/breathing apparatus.

**Additional information (precautions):** Move product containers away from fire or keep cool with water spray as a protective measure, where feasible. Use spark-proof tools and explosion-proof equipment. Avoid generating dust; fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a potential dust explosion hazard. Avoid inhaling gases, fumes, dust, mist, vapor, and aerosols. Avoid contact with skin, eyes, and clothing.

#### **SECTION 6 : Accidental release measures**

## Personal precautions, protective equipment and emergency procedures:

Wear protective equipment. Use spark-proof tools and explosion-proof equipment. Ensure that air-handling systems are operational. Ensure adequate ventilation.

## **Environmental precautions:**

Prevent from reaching drains, sewer or waterway. Collect contaminated soil for characterization per Section 13. Should not be released into environment.

#### Methods and material for containment and cleaning up:

Keep in suitable closed containers for disposal. Wear protective eyeware, gloves, and clothing. Refer to Section 8.Always obey local regulations. Dust deposits should not be allowed to accumulate on surfaces, as these may form an explosive mixture if they are released into the atmosphere in sufficient concentration. Avoid dispersal of dust in the air (i.e., clearing dust surfaces with compressed air). Collect solids in powder form using vacuum with (HEPA filter). Evacuate personnel to safe areas.

## **Reference to other sections:**

## SECTION 7: Handling and storage

## Precautions for safe handling:

Minimize dust generation and accumulation. Follow good hygiene procedures when handling chemical materials. Refer to Section 8.Do not eat, drink, smoke, or use personal products when handling chemical substances. Avoid contact with eyes, skin, and clothing.

## Conditions for safe storage, including any incompatibilities:

Store away from incompatible materials. Protect from freezing and physical damage. Keep away from food and beverages. Provide ventilation for containers. Avoid storage near extreme heat, ignition sources or open flame. Store in cool, dry conditions in well sealed containers. Store with like hazards

## SECTION 8: Exposure controls/personal protection

# according to 29CFR1910/1200 and GHS Rev. 3

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#### Calcium Metal,





**Control Parameters:** , , OSHA PEL TWA (Total Dust) 15 mg/m3 (50 mppcf\*) , , ACGIH TLV TWA (inhalable particles) 10 mg/m3

Appropriate Engineering controls: Emergency eye wash fountains and safety showers should be available in

the immediate vicinity of use/handling. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapor or dusts (total/respirable) below the applicable workplace exposure limits (Occupational Exposure Limits-OELs) indicated above. It is recommended that all dust control equipment such as local exhaust ventilation and material transport systems involved in handling of this product contain explosion relief vents or an explosion suppression system or an oxygen deficient environment. Ensure that dust-handling systems (such as exhaust ducts, dust collectors, vessels, and processing equipment) are designed in a manner to prevent the escape of dust into the work area (i.e., there is no leakage from the equipment). Use under a fume hood

Not required under normal conditions of use. Where risk assessment Respiratory protection:

> shows air-purifying respirators are appropriate use a full-face particle respirator type N100 (US) or type P3 (EN 143) respirator cartridges as a backup to engineering controls. When necessary use NIOSH approved

breathing equipment.

Protection of skin: Select glove material impermeable and resistant to the substance. Select

glove material based on rates of diffusion and degradation. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Use proper glove removal technique without touching outer surface. Avoid skin contact with used gloves. Wear

protective clothing.

Wear equipment for eye protection tested and approved under Eye protection:

appropriate government standards such as NIOSH (US) or EN 166(EU). Safety glasses or goggles are appropriate eye protection.

**General hygienic measures:** Perform routine housekeeping. Wash hands before breaks and at the end

of work. Avoid contact with skin, eyes, and clothing. Before wearing wash

contaminated clothing.

## SECTION 9: Physical and chemical properties

Appearance (physical state,color):	Silver white crystals	Explosion limit lower: Explosion limit upper:	Not determined Not determined
Odor:	Not Determined	Vapor pressure:	10 mm Hg @ 983 C
Odor threshold:	Not determined	Vapor density:	Not determined
pH-value:	Not Determined	Relative density:	Not determined
Melting/Freezing point:	845 C	Solubilities:	Decomposes
Boiling point/Boiling range:	1484 C	Partition coefficient (noctanol/water):	Not determined
Flash point (closed cup):	Not determined	Auto/Self-ignition temperature:	Not determined

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#### Calcium Metal,

Evaporation rate:	Not determined	Decomposition temperature:	Not determined
Flammability (solid,gaseous):	Not determined	Viscosity:	a. Kinematic:Not determined b. Dynamic: Not determined
Density: Not determined Specific Gravity:1.54@ 20C/4C			

## **SECTION 10: Stability and reactivity**

**Reactivity:**Water Reactive; material will react with water and may release a flammable and/or toxic gas **Chemical stability:**Combines vigorously or explosively with water. May ignite spontaneously if exposed to air or kept in the presence of moisture

**Possible hazardous reactions:** May ignite or explode on contact with steam or moist air; may re-ignite after fire is extinguished. This chemical poses an explosion hazard - reacts with water to form explosive hydrogen gas. None under normal processing

**Conditions to avoid:**Incompatible Materials.Dust generation, exposure to air, excess heat, exposure to moist air or water

**Incompatible materials:**Acids, oxidizing agents, carbonates, dinitrogen tetraoxide, halogens, sulfur, chlorine, chlorine trifluoride, fluorine, air, moisture, water, water and mixtures containing water (e.g. aqueous solutions, water).

**Hazardous decomposition products:**Irritating and toxic fumes and gases, hydrogen gas, calcium oxide, calcium hydroxide

## **SECTION 11: Toxicological information**

Acute Toxicity: No additional information.		
Chronic Toxicity: No additional information.		
Corrosion Irritation: No additional information.		
Sensitization:	No additional information.	
Single Target Organ (STOT):	No additional information.	
Numerical Measures:	No additional information.	
Carcinogenicity:	No additional information.	
Mutagenicity:	No additional information.	
Reproductive Toxicity:	No additional information.	

## **SECTION 12: Ecological information**

**Ecotoxicity Persistence and degradability:** 

**Bioaccumulative potential:** 

Mobility in soil:

Other adverse effects:

## **SECTION 13: Disposal considerations**

#### Waste disposal recommendations:

according to 29CFR1910/1200 and GHS Rev. 3

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#### Calcium Metal,

Contact a licensed professional waste disposal service to dispose of this material. Dispose of empty containers as unused product. Product or containers must not be disposed with household garbage. It is the responsibility of the waste generator to properly characterize all waste materials according to applicable regulatory entities (US 40CFR262.11). Chemical waste generators must determine whether a discarded chemical is classified as a hazardous waste. Chemical waste generators must also consult local, regional, and national hazardous waste regulations. Ensure complete and accurate classification.

## **SECTION 14: Transport information**

#### **UN-Number**

Not Regulated.

## **UN proper shipping name**

Not Regulated.

#### Transport hazard class(es)



#### Class

4.3 Substances which in contact with water emit flammable gases

Packing group: Not Regulated Environmental hazard: Transport in bulk:

Special precautions for user:

## **SECTION 15: Regulatory information**

#### **United States (USA)**

## SARA Section 311/312 (Specific toxic chemical listings):

Reactive, Acute, Chronic

## SARA Section 313 (Specific toxic chemical listings):

None of the ingredients is listed

#### RCRA (hazardous waste code):

None of the ingredients is listed

#### TSCA (Toxic Substances Control Act):

All ingredients are listed.

## CERCLA (Comprehensive Environmental Response, Compensation, and Liability Act):

None of the ingredients is listed

## Proposition 65 (California):

## Chemicals known to cause cancer:

None of the ingredients is listed

## Chemicals known to cause reproductive toxicity for females:

None of the ingredients is listed

## Chemicals known to cause reproductive toxicity for males:

None of the ingredients is listed

#### Chemicals known to cause developmental toxicity:

None of the ingredients is listed

#### Canada

according to 29CFR1910/1200 and GHS Rev. 3

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#### Calcium Metal,

#### Canadian Domestic Substances List (DSL):

All ingredients are listed.

#### Canadian NPRI Ingredient Disclosure list (limit 0.1%):

None of the ingredients is listed

## Canadian NPRI Ingredient Disclosure list (limit 1%):

None of the ingredients is listed

#### **SECTION 16: Other information**

This product has been classified in accordance with hazard criteria of the Controlled Products Regulations and the SDS contains all the information required by the Controlled Products Regulations.Note:. The responsibility to provide a safe workplace remains with the user.The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment.The information contained herein is, to the best of our knowledge and belief, accurate.However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material.It is the responsibility of the user to comply with all applicable laws and regulations applicable to this material.

#### **GHS Full Text Phrases:**

## **Abbreviations and acronyms:**

IMDG: International Maritime Code for Dangerous Goods

IATA: International Air Transport Association

GHS: Globally Harmonized System of Classification and Labelling of Chemicals

ACGIH: American Conference of Governmental Industrial Hygienists

CAS: Chemical Abstracts Service (division of the American Chemical Society)

NFPA: National Fire Protection Association (USA) HMIS: Hazardous Materials Identification System (USA)

WHMIS: Workplace Hazardous Materials Information System (Canada)

DNEL: Derived No-Effect Level (REACH)

PNEC: Predicted No-Effect Concentration (REACH)

CFR: Code of Federal Regulations (USA)

SARA: Superfund Amendments and Reauthorization Act (USA)

RCRA: Resource Conservation and Recovery Act (USA)

TSCA: Toxic Substances Control Act (USA)

NPRI: National Pollutant Release Inventory (Canada)

DOT: US Department of Transportation

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